

Understanding Inherent Substrate Selectivity During Tungsten ALD

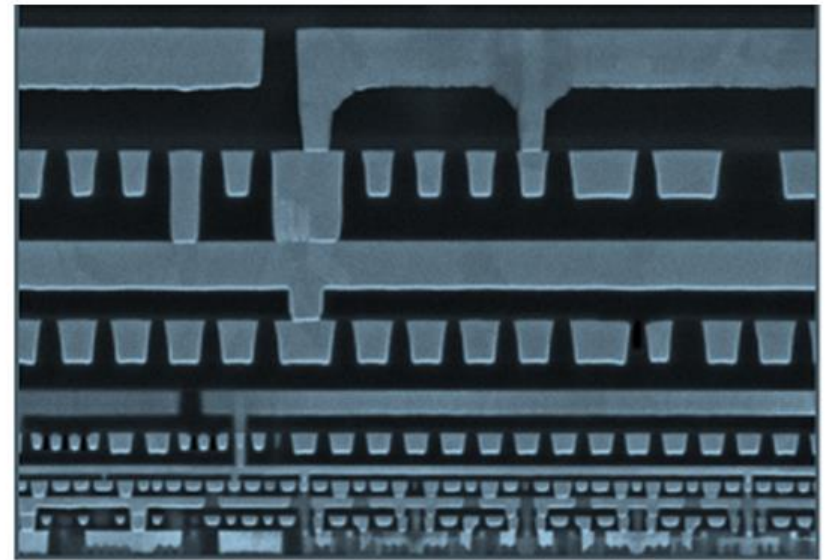
Paul Lemaire, Mariah King, Gregory Parsons
Dept. of Chem.& Biomol. Eng., North Carolina State University
Raleigh, NC 27695, USA

2nd Area Selective Deposition Workshop ASD17

Introduction

- Transistor devices are becoming smaller with more complex architecture
- Conventional patterning processes are limited:
 - Maintaining edge definition and alignment to underlying features.
 - Small dimensions,
 - High aspect ratio features
 - Use of ultra-low k dielectrics
- Need for selective deposition
 - Blocking layers (SAMs)
 - Competitive adsorption
 - Selective etching
 - **Inherent selectivity**

14 nm Process

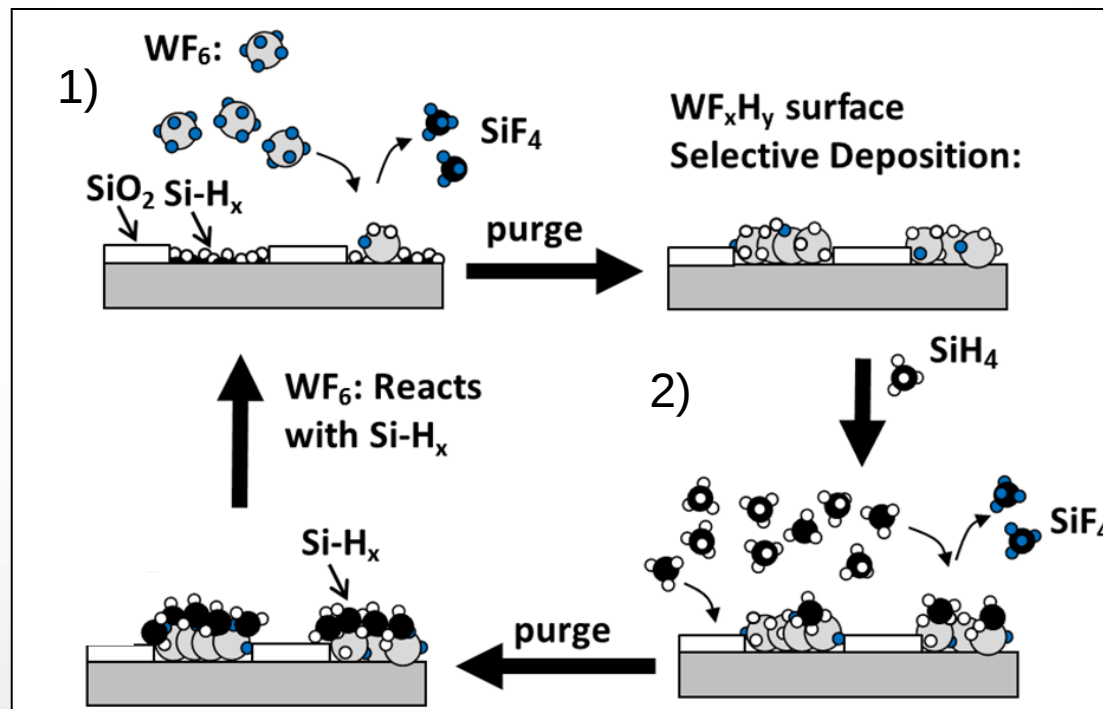


52 nm (0.65x) minimum pitch

Source: Intel (<http://www.anandtech.com>)

Background: Selective W-ALD

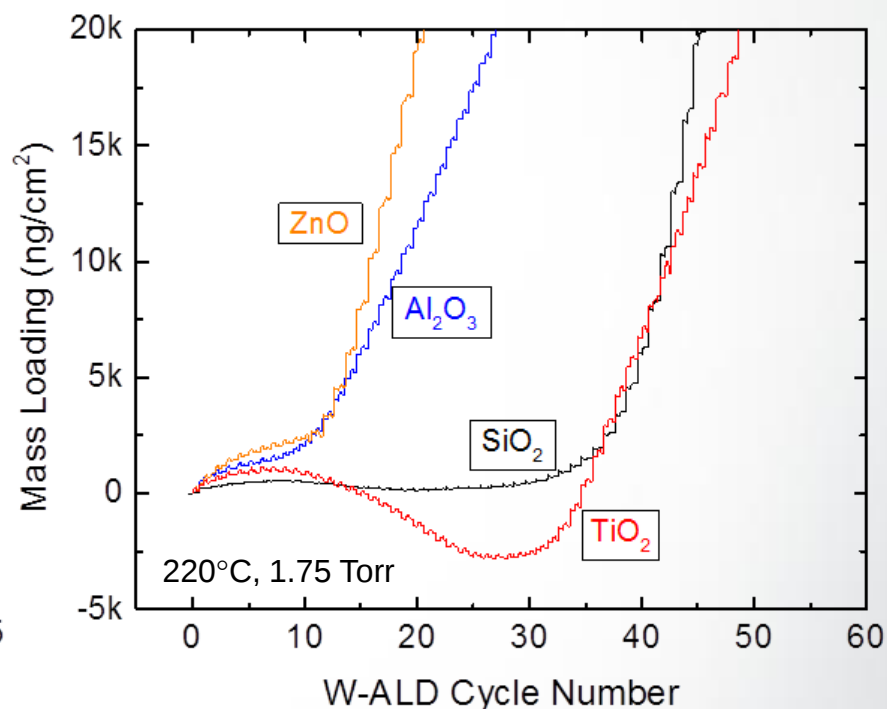
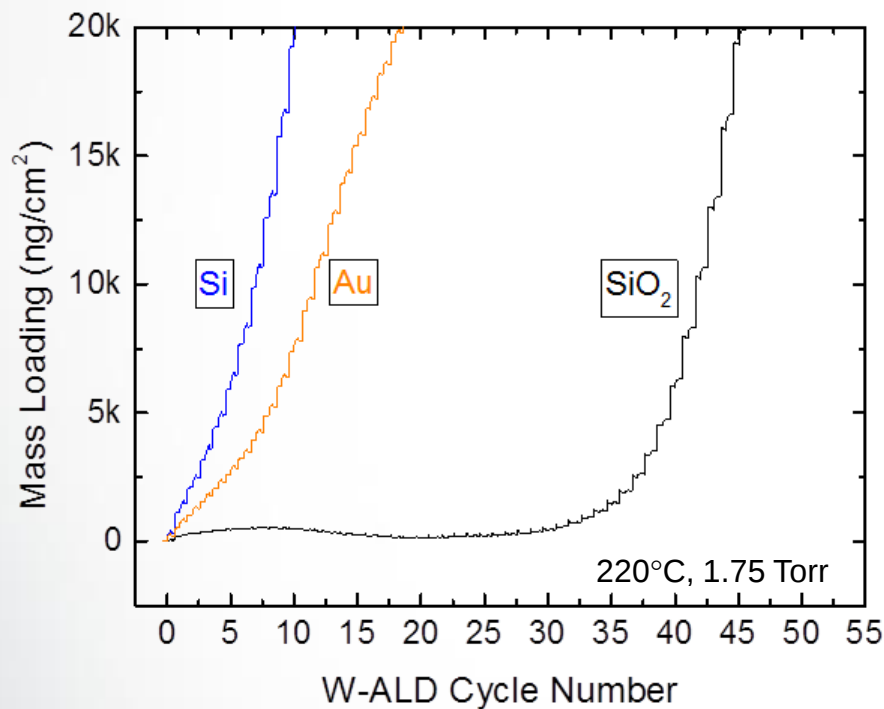
- **Atomic Layer Deposition (ALD):** deposit conformal films with high thickness control with vapor-phase, self-limiting reactions
- Deposit tungsten with tungsten hexafluoride (WF_6) and dilute silane (SiH_4) at low temperatures (220°C)



Inherently Selective W-ALD

- Need to determine:
 1. Substrate surface defects that lead to nucleation
 2. Reactant that initiates undesired deposition
- Study phenomena that either inhibit or promote deposition
 1. Generate and remove nucleation sites via sample prep
 2. Pre-expose substrates to W-ALD precursors
- Improve selectivity by varying temperature and concentration

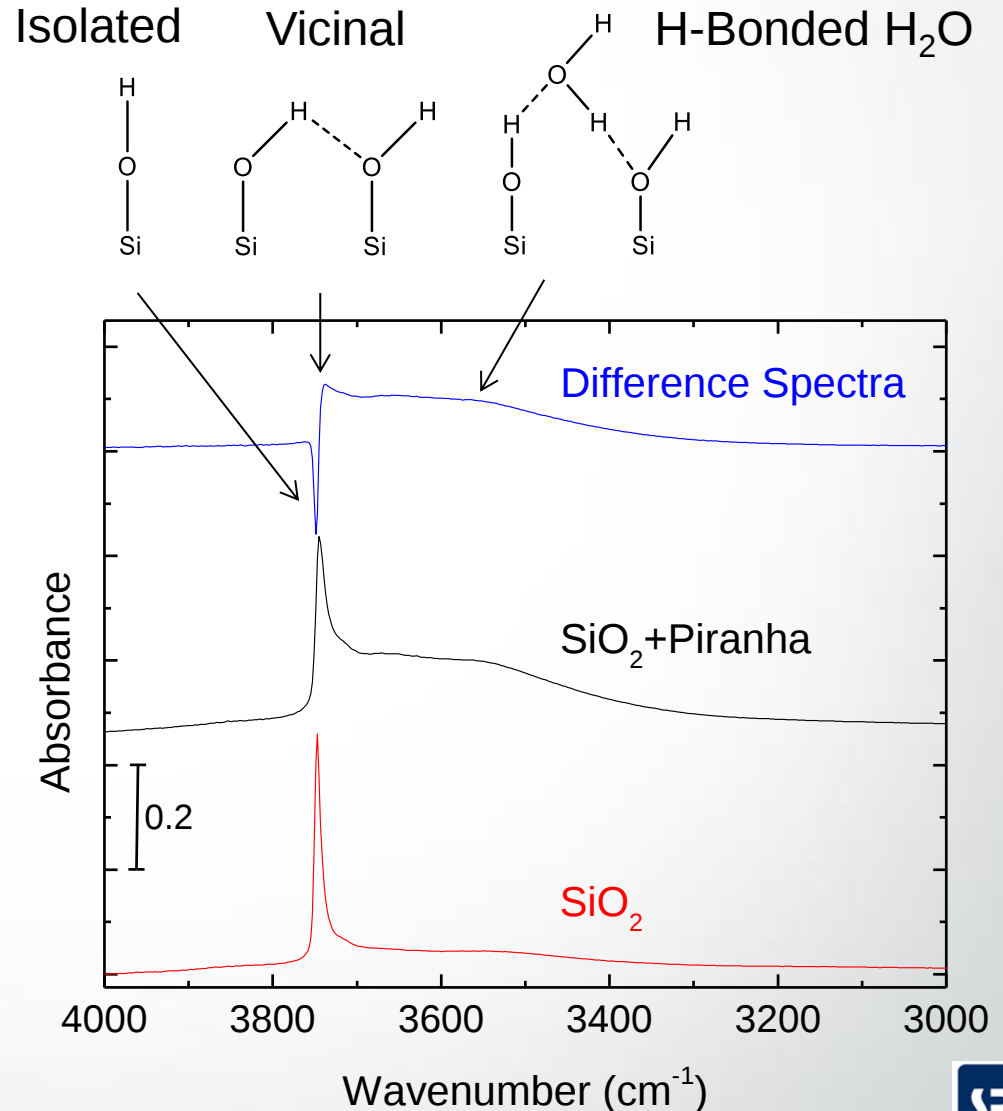
W-ALD Nucleation (QCM)



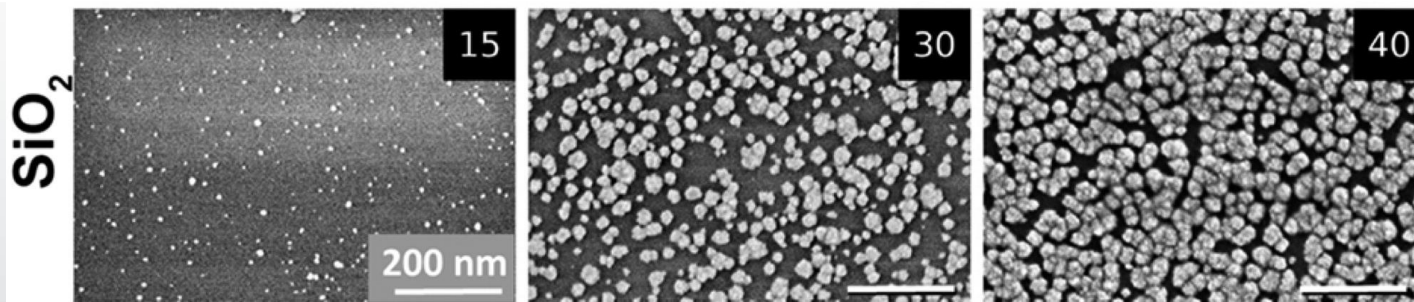
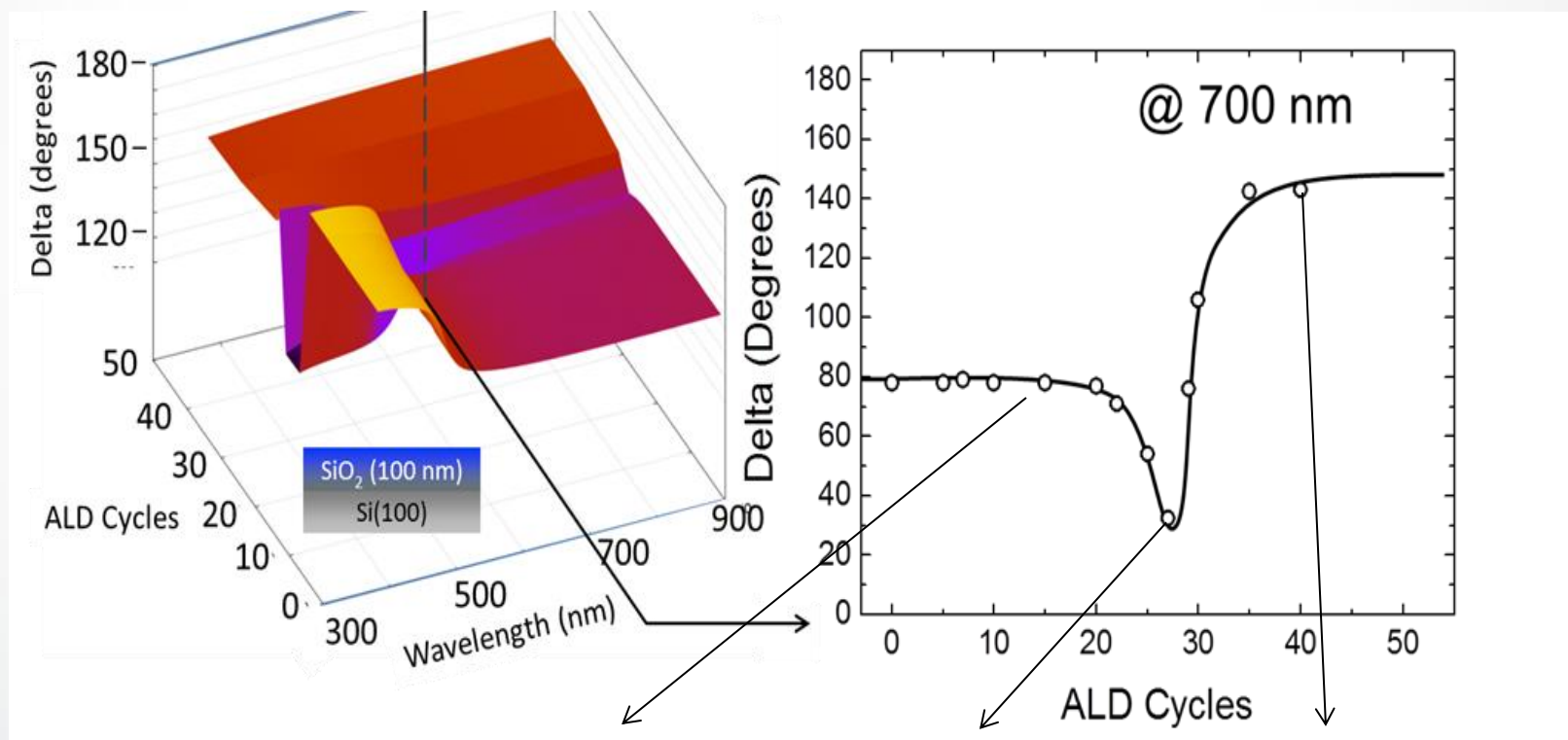
- Little to no tungsten nucleation delay on Si and Au
 - Selectively deposit ~15 nm of W on Si or Au
- Tungsten nucleation delay dependent on oxide composition
- TiO₂ etching during nucleation period

Hydroxyls and Sample Preparation (FTIR)

- Porous SiO_2 substrates
 - Anodization of Si in HF/EtOH
 - Thermal oxidation at 950°C in air
- As-is SiO_2 :
 - strong isolated OH peak
- Piranha cleaned SiO_2 :
 - Loss of isolated OH
 - generate vicinal OH and H-bonded H_2O

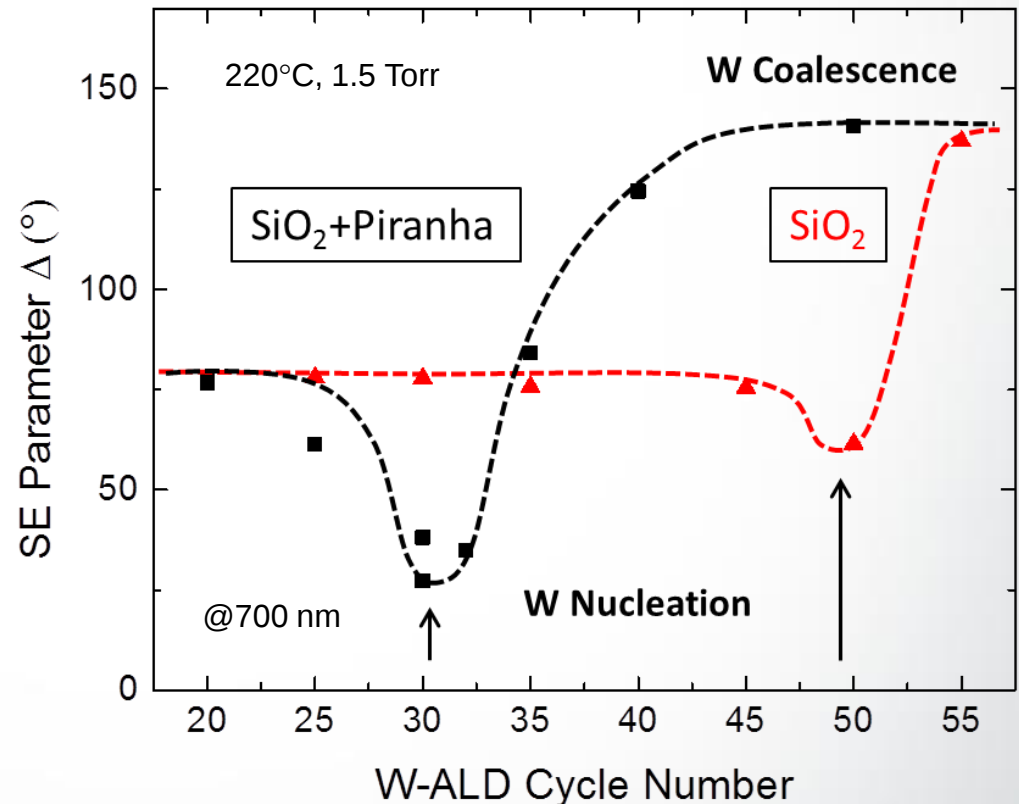


Ellipsometric Analysis of W-ALD



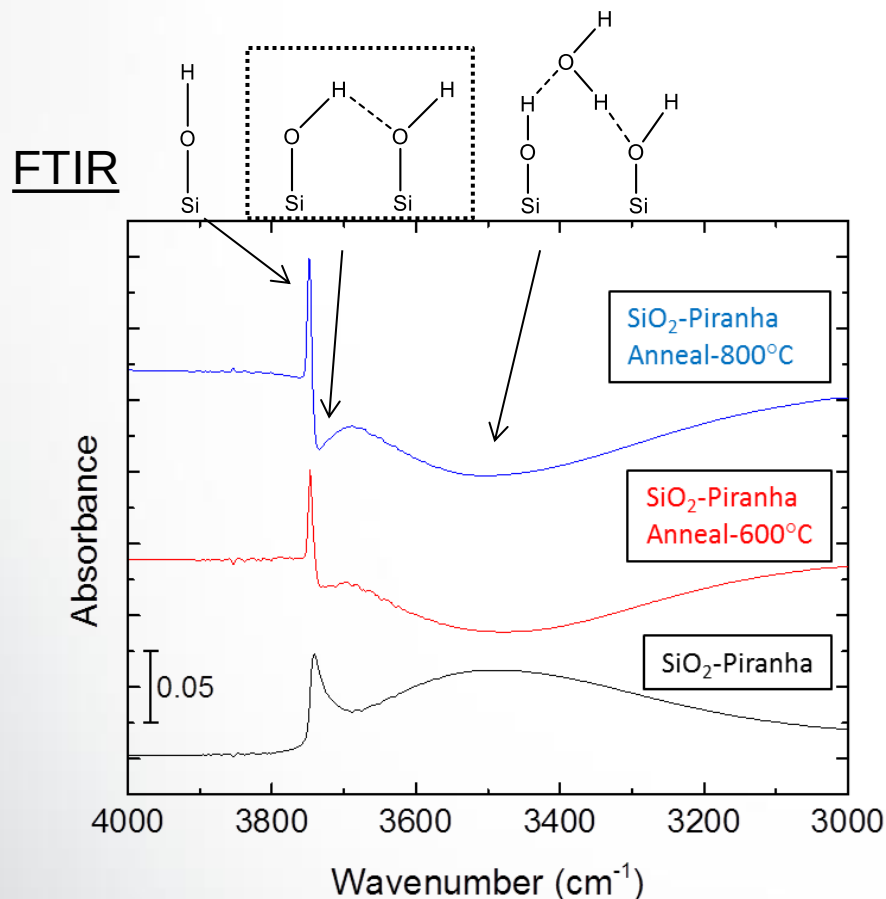
Sample Preparation (Ellipsometry)

- **Piranha cleaned SiO_2**
 - Increase in vicinal OH
 - H-bonded water
 - Nucleation: 35 cycles
- **As-is SiO_2**
 - Isolated OH
 - Nucleation: 50 cycles

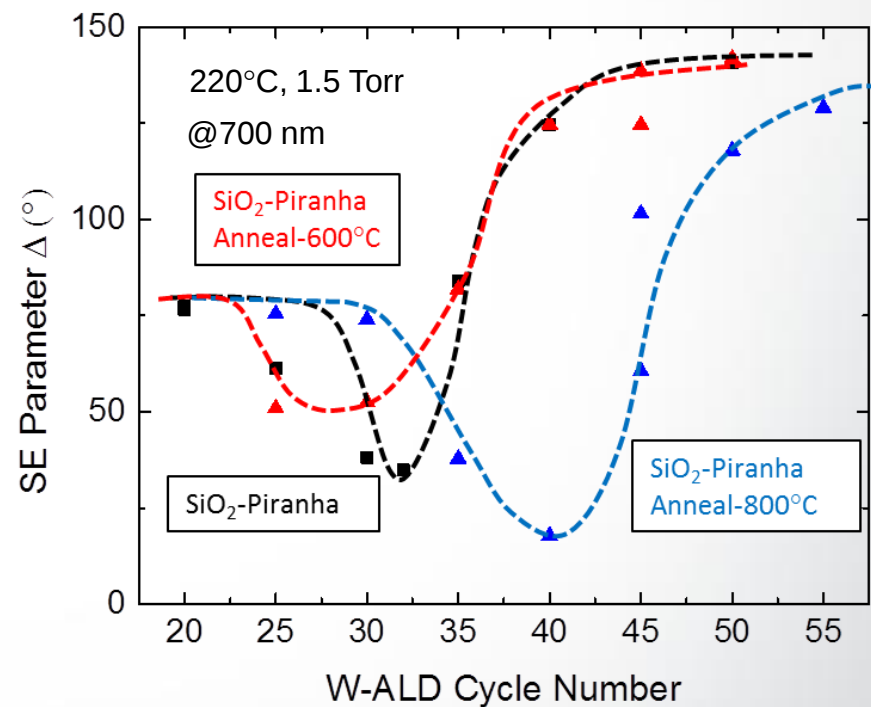


Piranha clean promotes W nucleation by forming vicinal OH sites

Sample Preparation: Removing OH Sites

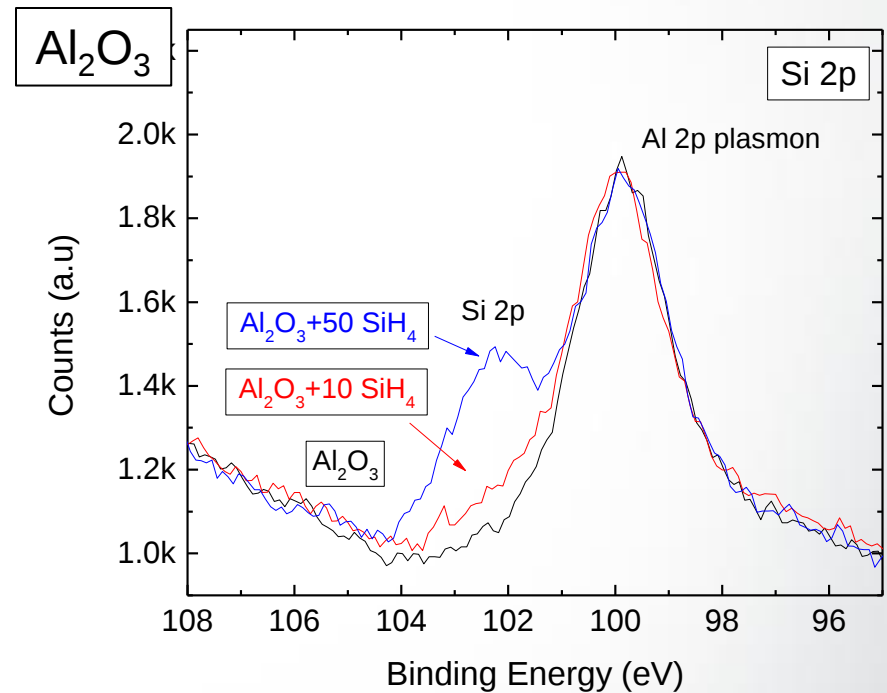
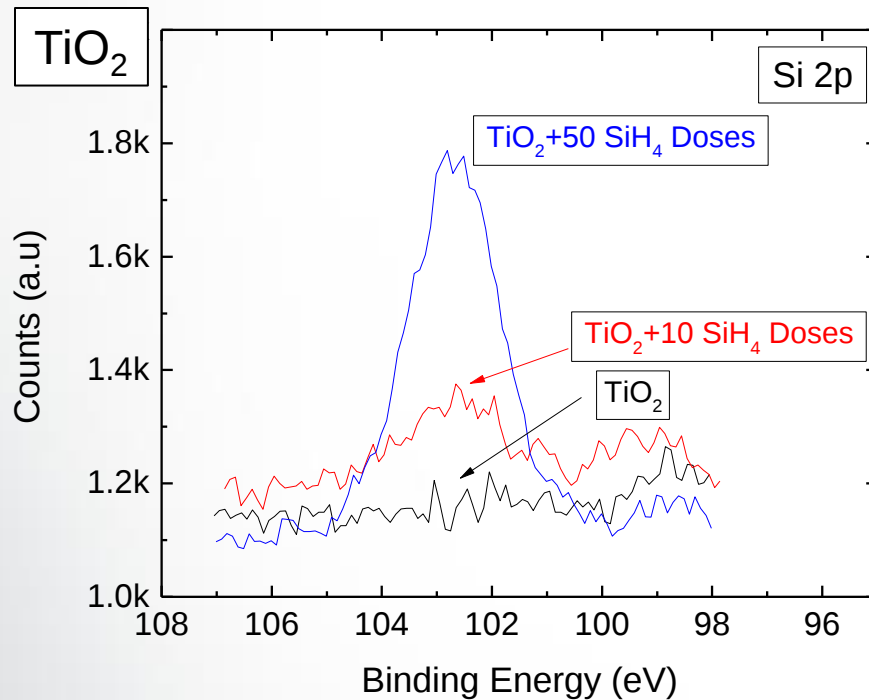


Ellipsometry



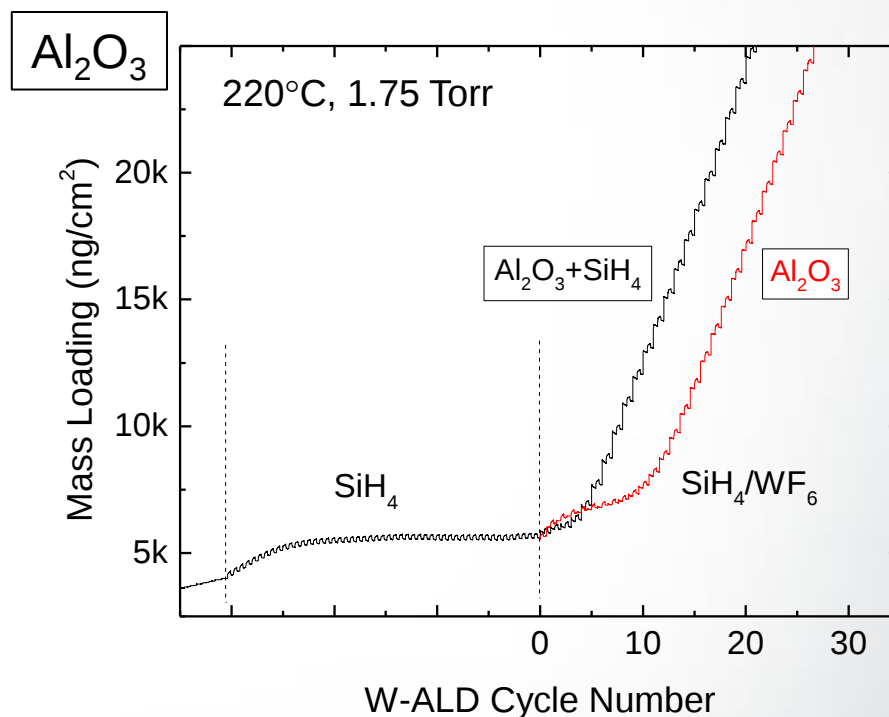
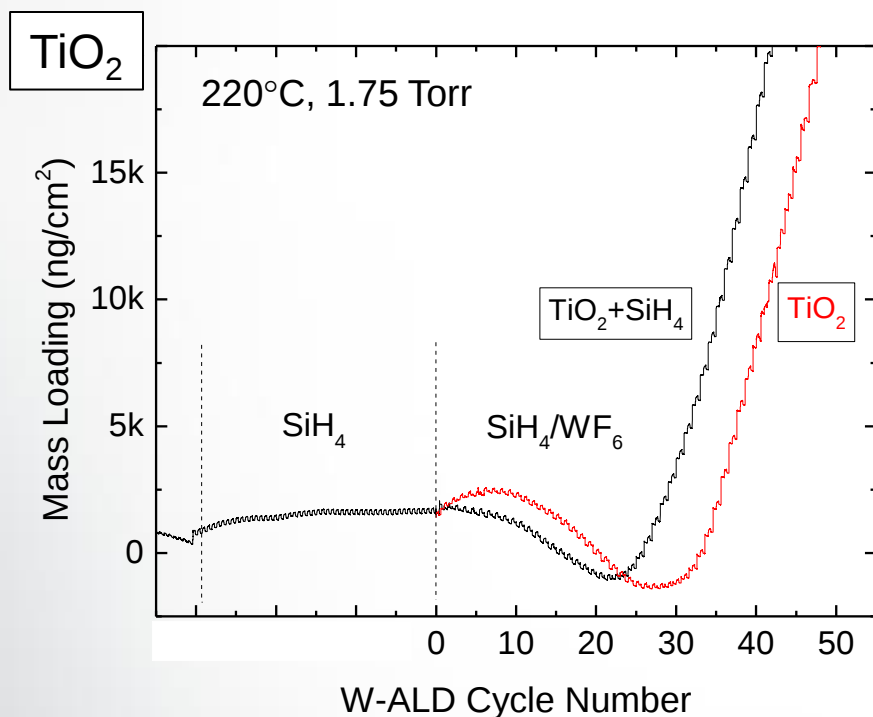
- Tungsten nucleation occurs predominantly at vicinal hydroxyl sites

$\text{TiO}_2/\text{Al}_2\text{O}_3 + 1) \text{SiH}_4$ (XPS)



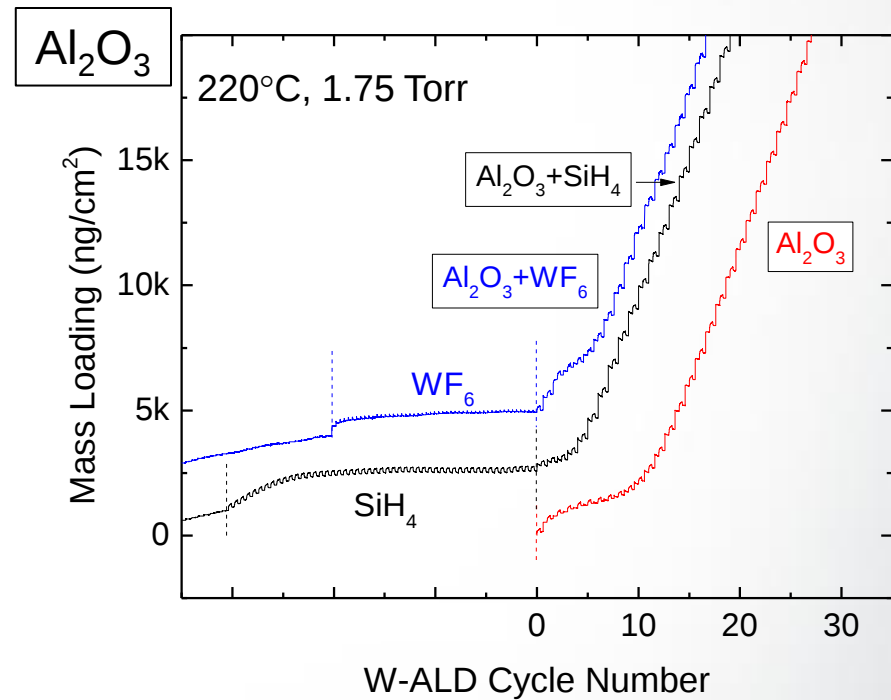
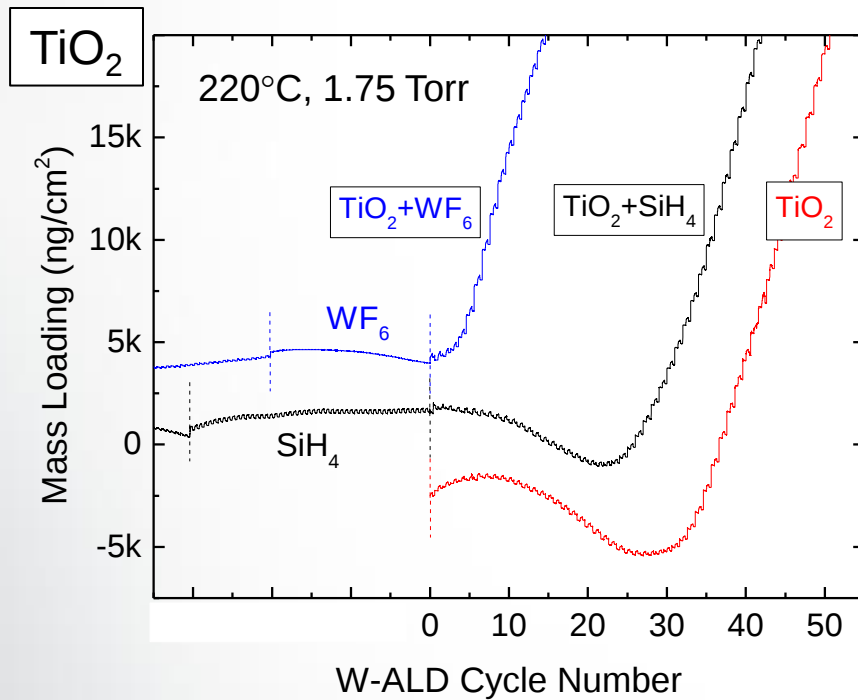
- Increase in Si 2p signal on oxide substrate with SiH₄ exposures
 - Binding energy consistent with oxidized Si

$\text{TiO}_2/\text{Al}_2\text{O}_3 + 1) \text{SiH}_4 + 2) \text{W-ALD (QCM)}$



- In-situ QCM assessment of SiH_4 pre-exposures on TiO_2 and Al_2O_3
 - 50 SiH_4 pre-exposures: tungsten nucleation occurs ~10 cycles earlier

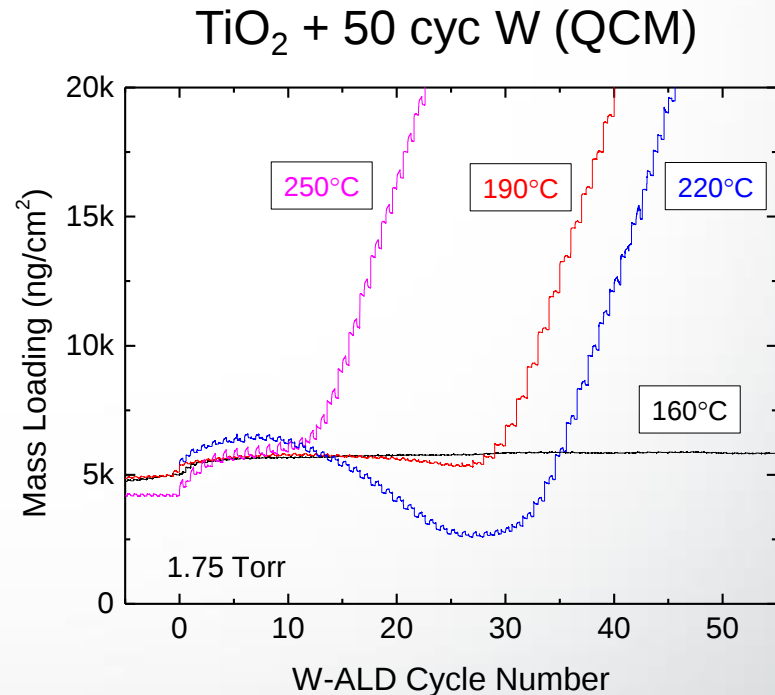
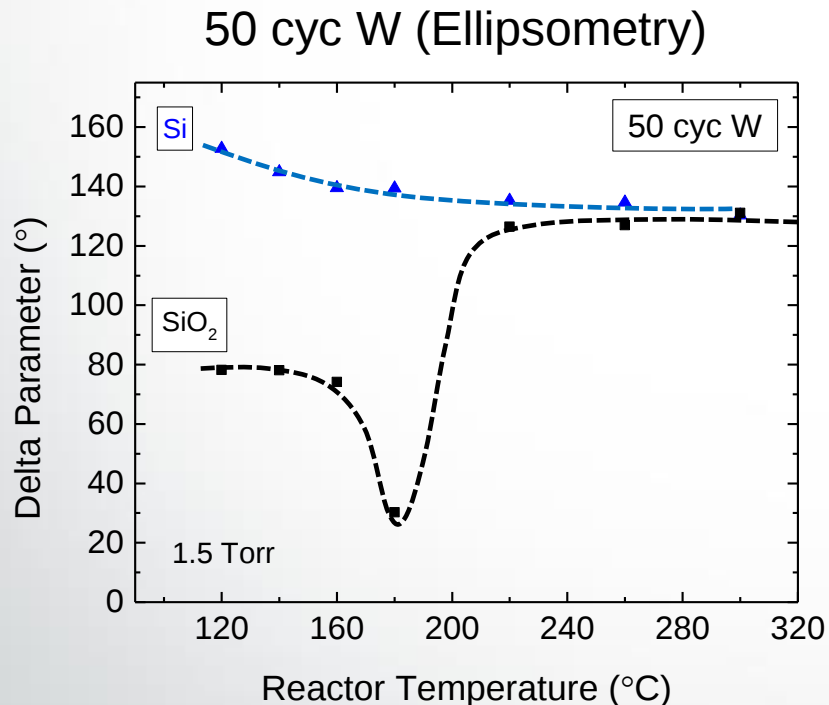
$\text{TiO}_2/\text{Al}_2\text{O}_3$ + 1) WF_6 + 2) W-ALD (QCM)



- Minimal nucleation delay on $\text{Al}_2\text{O}_3/\text{TiO}_2$ following 50 WF_6 pre-exposures
- Tungsten nucleation more dependent on WF_6 than SiH_4 pre-exposures

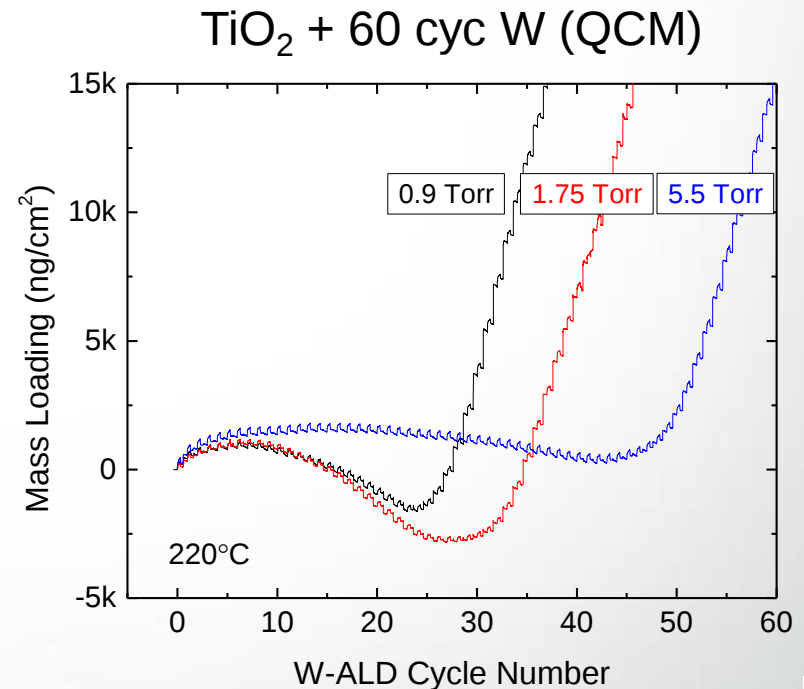
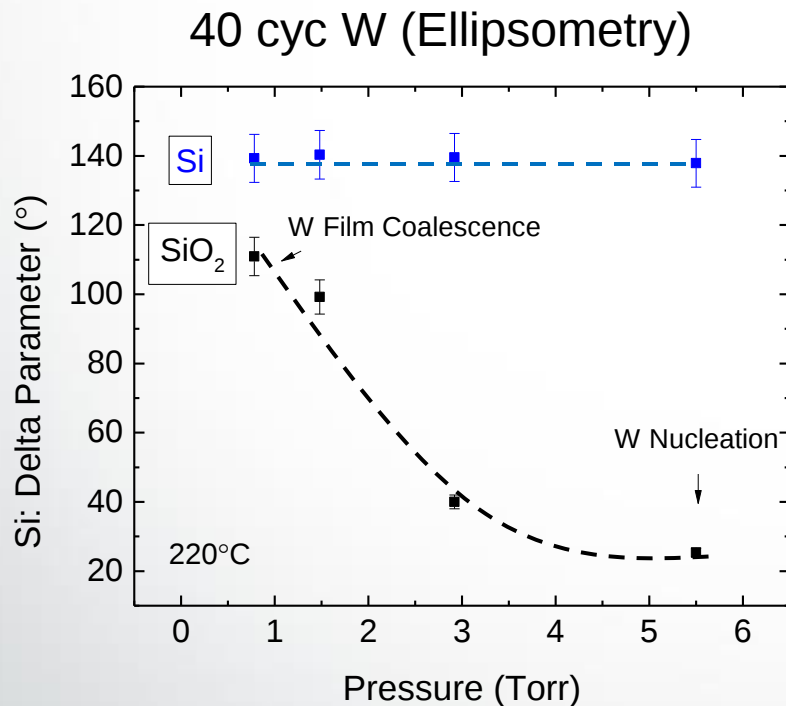
W-ALD Selectivity: Deposition Temperature

- Reduce tungsten nucleation on SiO_2 and TiO_2 at lower temperatures
 - Unable to overcome activation barrier for adsorption or reaction
 - Also reduces desired W deposition on Si
- Etching occurs in narrow temperature range



W-ALD Selectivity: Process Pressure/Flowrate

- Tungsten nucleation on oxides delayed at higher carrier flowrate/system pressure
- Lower precursor concentration limits precursor adsorption on oxides

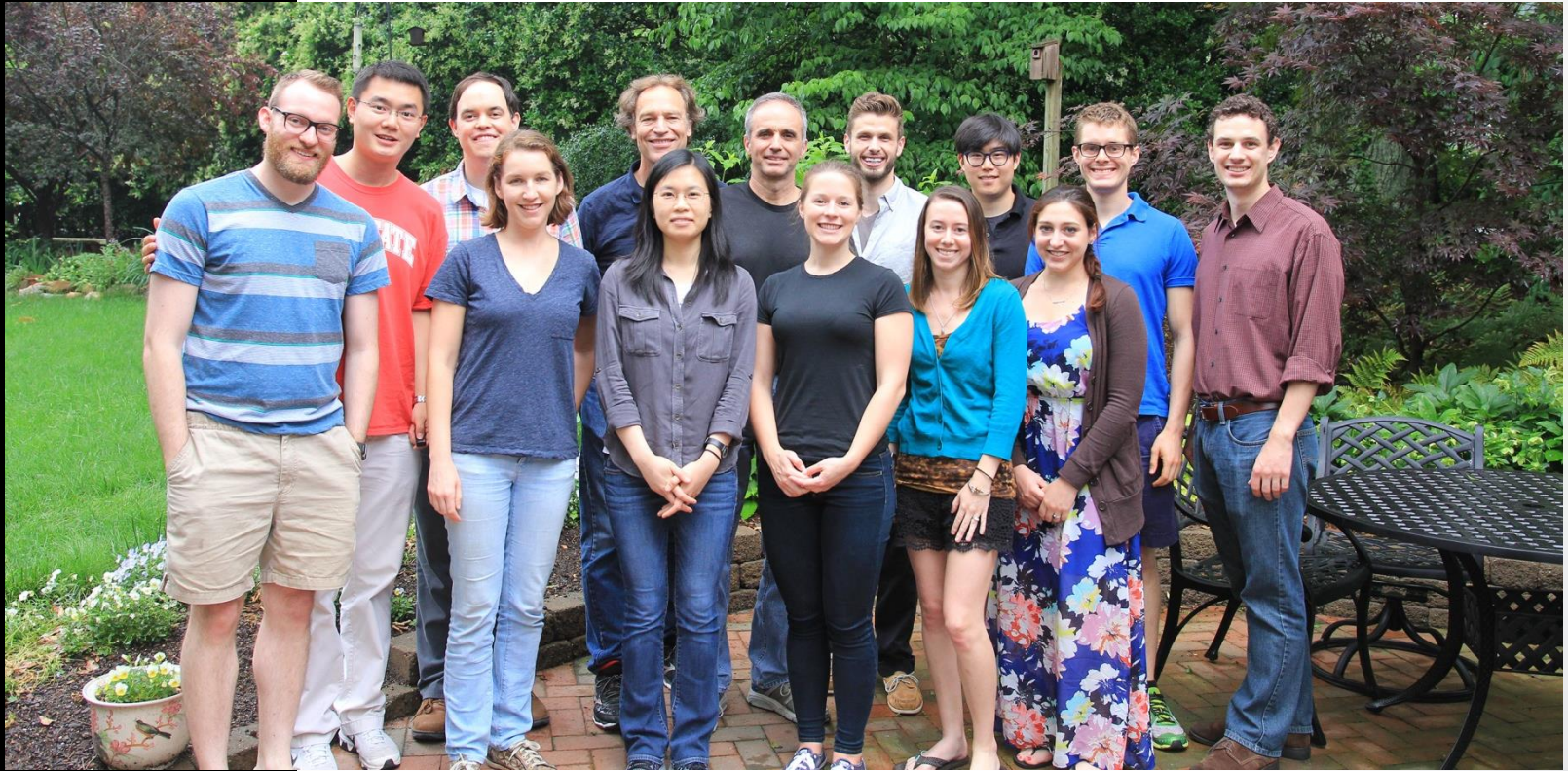


Summary

- Tungsten nucleation delay is substrate dependent:
 $\text{Si} > \text{Au} > \text{Al}_2\text{O}_3/\text{ZnO} > \text{TiO}_2/\text{SiO}_2$
- Correlation between vicinal hydroxy concentration and W-ALD nucleation
- Surface bound water inhibits W-ALD nucleation
- SiH_4 pre-exposures promote nucleation by forming reducing layer
- Reduce temperature or increase flow rate to limit WF_6 adsorption on oxide surfaces

Lemaire, P. C.; King, M.; Parsons, G. N.
J. Chem. Phys. **2017**.

Acknowledgements



- Gregory N Parsons Research Group
- Funding from Semiconductor Research Corporation (SRC)