

# mec

# AREA SELECTIVE Ru ALD FOR SUB 7 NM BOTTOM-UP METAL INTERCONNECTS

I. Zyulkov<sup>1,2</sup>, M. Krishtab<sup>1,2</sup>, S. De Gendt<sup>1,2</sup>, S. Armini<sup>2</sup>

<sup>1</sup>KU Leuven, Faculty of Science, B-3001 Leuven, Belgium

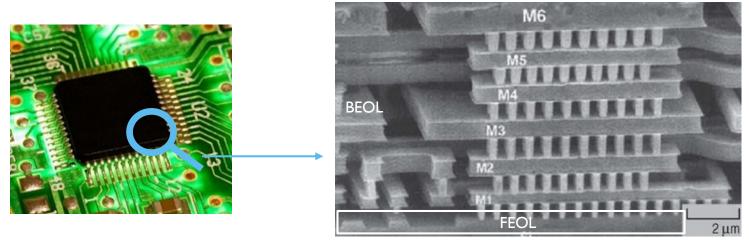
<sup>2</sup> Imec, Kapeldreef 75, B-3001 Leuven, Belgium

# **OUTLINE**

- Introduction and Motivation
- Process flow
- Process fundamentals
  - Plasma interaction with SiCN/ Low-k and aC surfaces
  - Ru selectivity vs plasma treatments
  - Ru selectivity vs ALD temperature
- ASD Cu/ Ru integration (45 nm HP)
  - Test vehicle
  - Ru selectivity vs plasma treatments
  - Defectivity
- Conclusions and outlook



# What are Interconnects?



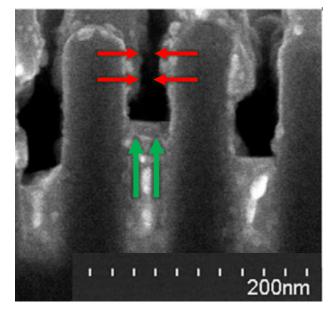
Multilevel metallization structure

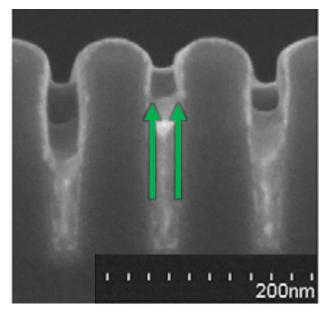
- IC consists of devices (FEOL) and interconnects (BEOL)
- Interconnects consist of metal and dielectric isolation

innec

3

# Problem statement



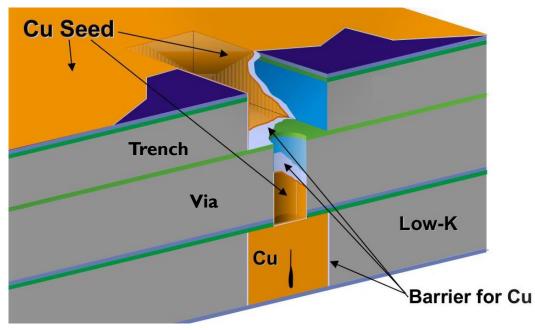


iPVD Cu deposition without (left) and with (right) Ru liner, Ishizaka, Tadahiro, et al., Microelectronic Engineering 92 (2012): 76-78

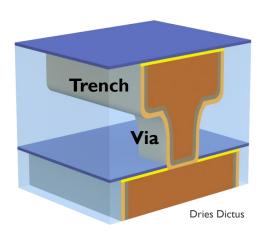
- Conventional metal fill becomes very challenging for narrow structures
- Bottom-up metal deposition can be used for defect-free fill

innec

# Interconnect structure. Dual-Damascene



# Dual-Damascene



Multilevel metallization structure

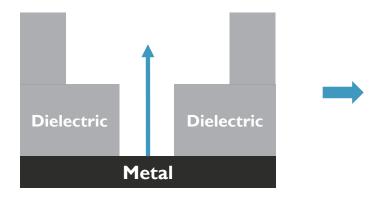
- Metal fill is used instead of metal etch
- Metal fill of Via and Trench areas is required (bottom-up?)

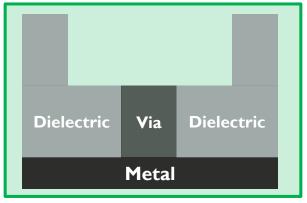
່ເກາຍເ

5

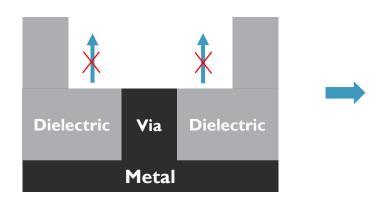
# Electroless deposition (ELD) in DD

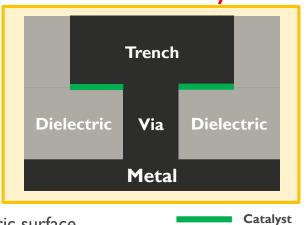
# **S**electivity by the nature





#### **Selective catalyst is needed**



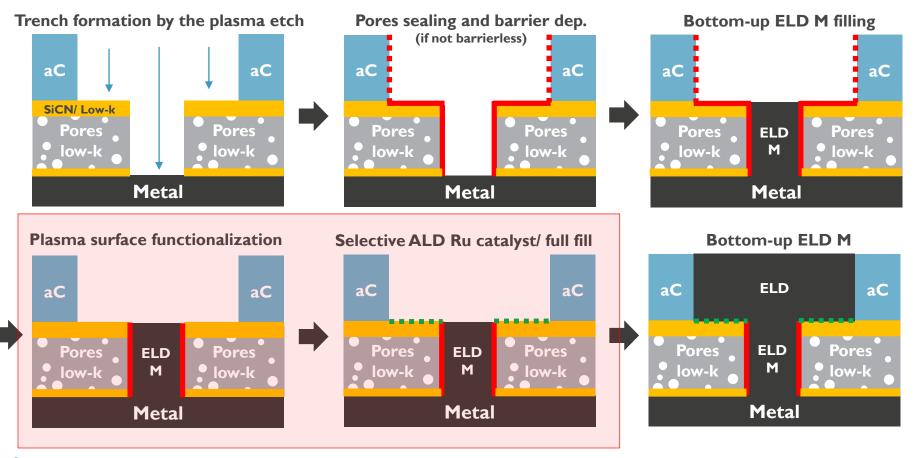


Metal catalyst is needed for Electroless deposition on the dielectric surface

ımec

# Process flow

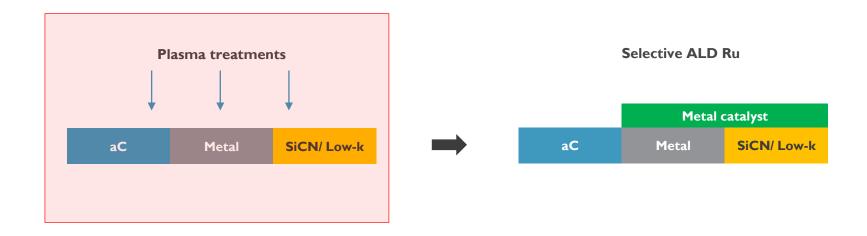
# Trench-up deposition in alternative DD structure using plasma for surface functionalization



unec

aC etch and low-k replacement

# Process scheme



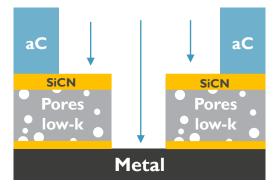
Plasma treatment is the most important step to achieve ALD Ru selectivity

imec

# Plasma interaction with SiCN/ Low-k and aC surfaces

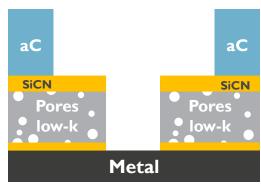
# Plasma needed for the process

#### Trench formation by the plasma etch



 $H_2/N_2$  plasma (provides anisotropic etch of dielectrics)

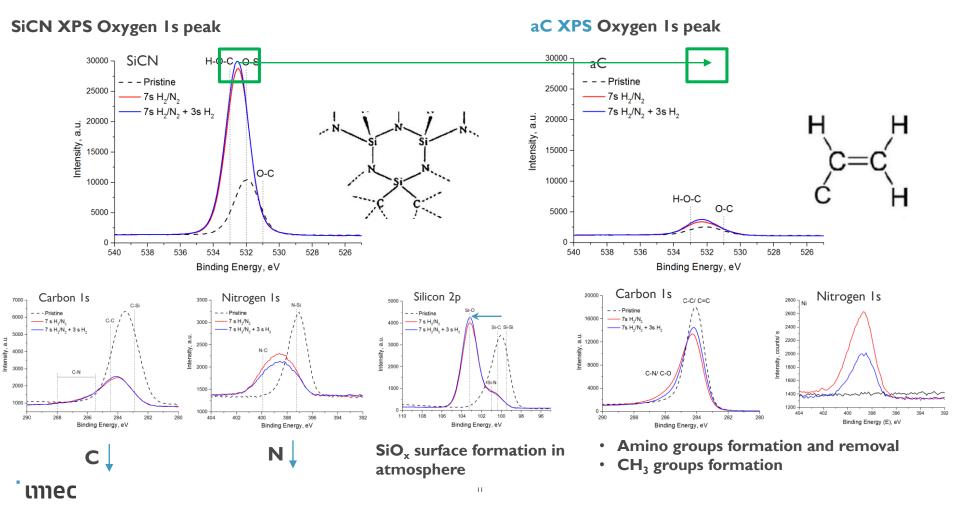
#### Plasma surface functionalization



Different plasma mixtures were investigated such as: CF<sub>4</sub>/H<sub>2</sub>, CH<sub>4</sub>/H<sub>2</sub>, H<sub>2</sub>.

Selectivity was achieved with H<sub>2</sub> plasma post-treatment

# Plasma interaction with dielectric surfaces

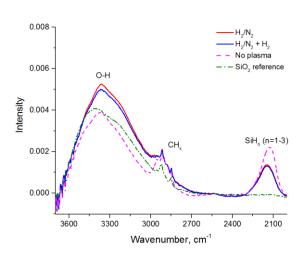


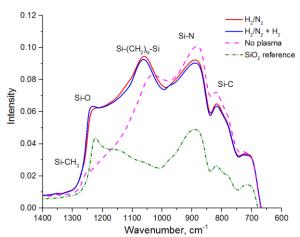
# Plasma interaction with dielectric surfaces

#### Water contact angle measurements:

# SiCN WCA, degrees ac WCA, degrees ac WCA, degrees ac WCA, degrees

#### FTIR spectra of SiCN before/ after plasma treatments

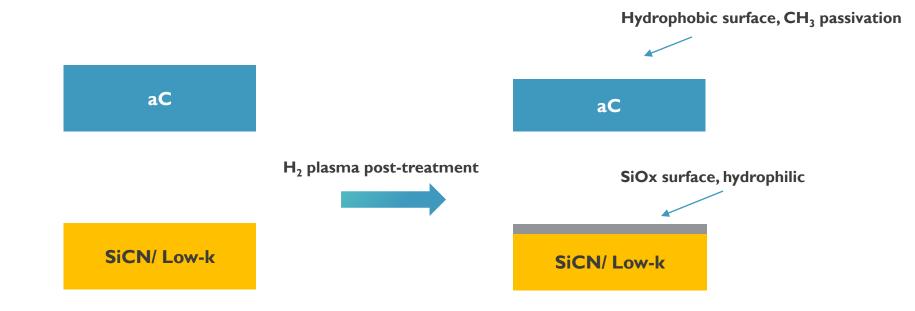




- SiCN becomes more hydrophilic after the plasma treatment
- aC properties restores after H<sub>2</sub> plasma



# Plasma interaction with dielectric surfaces

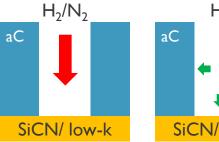


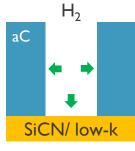


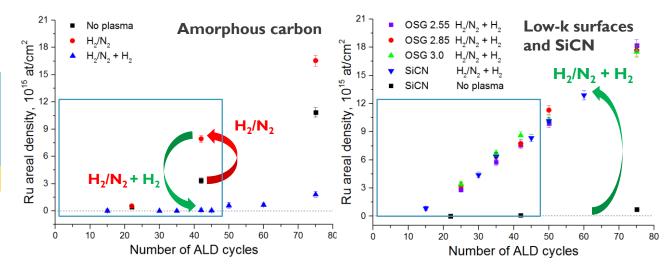
Selective ALD Ru for bottom-up metal ELD

# ALD Ru selectivity vs plasma treatment

#### 2 plasma steps in the process flow

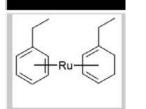






# Ethylbenzen-ethylcyclohexadiene Ru(0) precursor

#### EBECHRu



Precursor decomposition occurs at 225-325 C in the presence of oxygen



# Selectivity window up to 45 cycles (~ I nm Ru on SiCN) @ 325 C, sel. factor 400

Heo, Jaeyeong, et al., Electrochemical and Solid-State Letters 11.2 (2008), G5-G8.

"surface modification by the UV-assisted O3 treatment, causing it to be denser and hydrophilic, i.e., an oxygenrich composition. The increase in the number of reactive sites led to the easy chemisorption and cracking of the Ru precursor by the sufficient supply of oxygen"

Heo, Jaeyeong, et al., Electrochemical and Solid-State Letters 11.8 (2008), H210-H213.

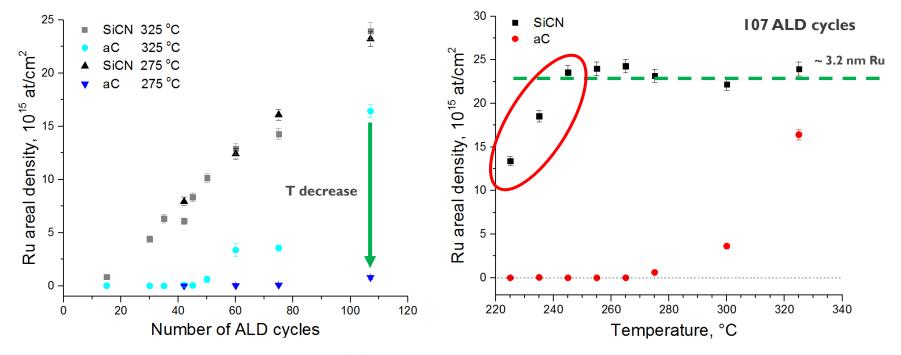
"a weak Van der Waals force may be the only interacting force between the ligand and the methyl group [..] it has been reported that the aromatic benzene molecule ( $C_6H_6$ ) exhibits a strong adsorption on a silanol group.."

innec

5

# ALD Ru selectivity vs deposition Temperature

RBS Ru concentrations on SiCN/ aC surface



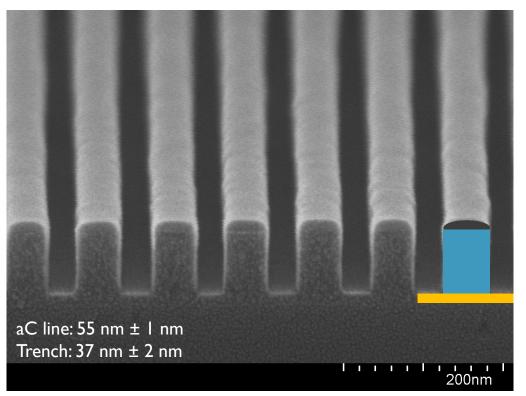
- Two deposition regimes are observed on SiCN
- Ru GR on aC constantly decreases on aC surface while Ru GR is constant on SiCN up to 250 C
- Selectivity window up to 3.2 nm Ru on SiCN (sel. factor 600) @ 275 C

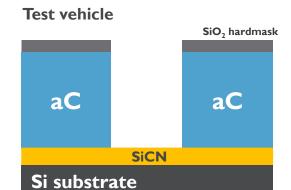


ASD Ru/ Cu on patterned substrates

# Test vehicle (45 nm HP)

 $H_2/N_2$  plasma etch of aC/ SiCN





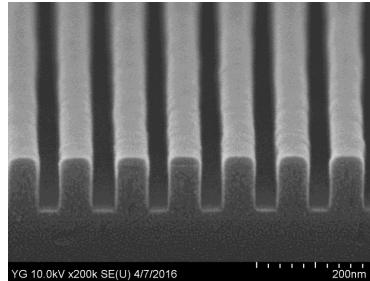
- ← 20 nm SiO<sub>2</sub> hardmask
- **← 80 nm PECVD aC**
- ← 15 nm PECVD SiCN

H<sub>2</sub>/N<sub>2</sub> plasma is very anisotropic and selective to aC with respect to SiCN

innec

# Test vehicle (45 nm HP)

# H<sub>2</sub>/N<sub>2</sub> plasma etch of aC/ SiCN:



HF clean +

3 s H<sub>2</sub> plasma



HF clean +

10 s H<sub>2</sub> plasma



16 10.0KV X200K 3E(0) 4/1

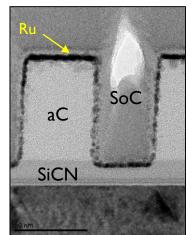
- Short H<sub>2</sub> plasma (3 s)
  ALD Ru is not selective
- N is detected on aC sidewalls and aC lines top

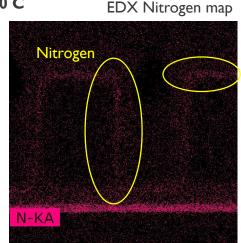
Long H<sub>2</sub> plasma (10 s)

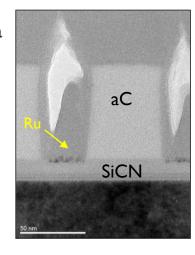
- ALD Ru is selective at the trench bottom
- N content is close to the noise signal

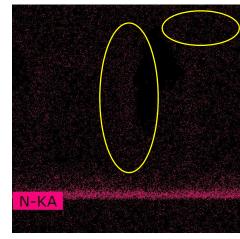
imec











## Electroless Cu on ALD Ru seed

Model ELD Cu chemistry

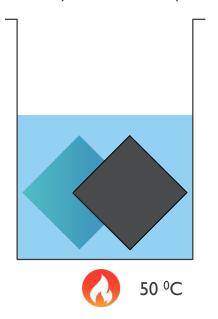
At 50 °C in a glovebox (concentration of oxygen did not exceed 10 ppm)

The ELD solution contains:

- 0.036 mol/L Cu sulfate
- 0.24 mol/L ethylenediaminetetraacetic acid (EDTA) as a complexing agent
- 40 ppm of 2-2 dipyridyl as a stabilizer
- 0.19 mol/L glyoxylic acid as a reducing agent
- 500 ppm of polyethylene glycol (PEG Mw 4000)

The pH of the bath was adjusted by tetramethylammonium hydroxide (TMAH; Fujifilm, Silicon Etch)

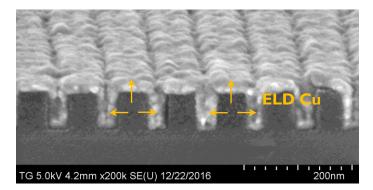
Coupons were dipped in the ELD bath simultaneously for the same deposition time

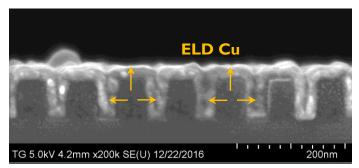


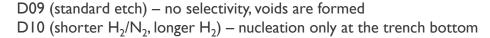


# ELD Cu (5 min)

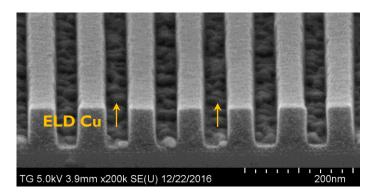
#### Short H<sub>2</sub> plasma

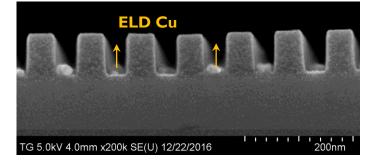






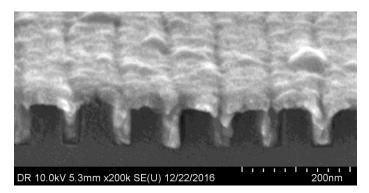
# Long H<sub>2</sub> plasma

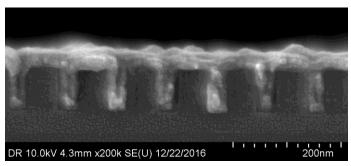




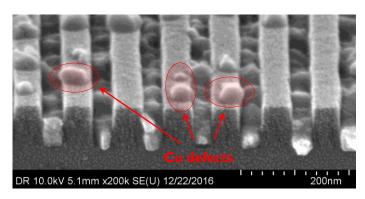
# ELD Cu (10 min)

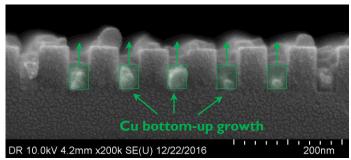
## Short H<sub>2</sub> plasma





## Long H<sub>2</sub> plasma



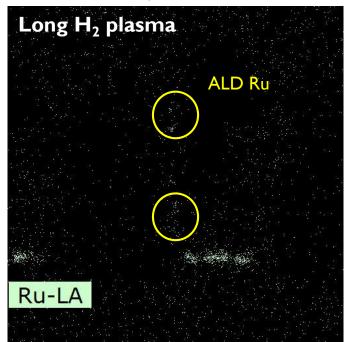


Cu defects are formed on aC lines ELD or ALD defects?

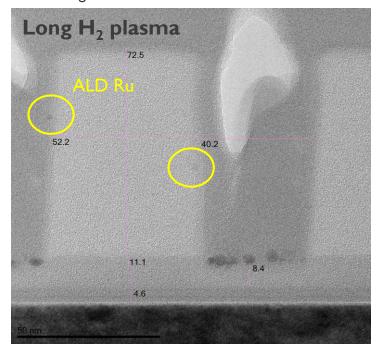


# **Defectivity**

EDX Ruthenium map



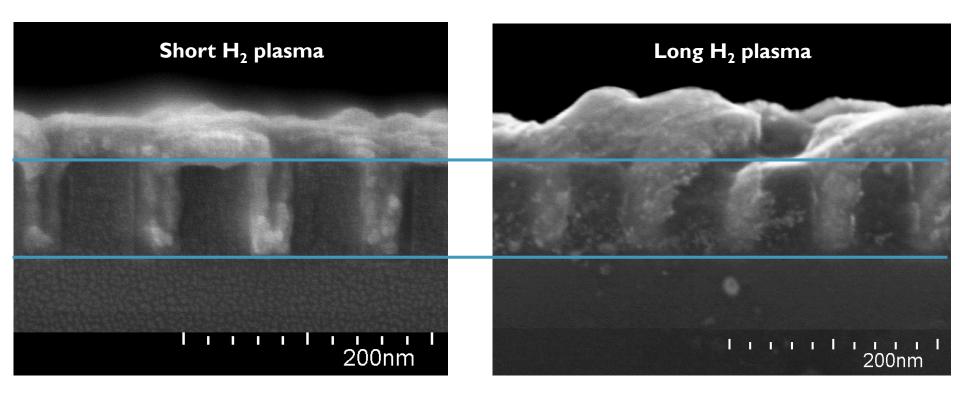
TEM image



- There are Ru defects on aC (both sidewalls and lines top).
- ALD Ru seed is not uniform at the bottom of the trench.



# ELD Cu full fill



No central seam is formed in the case of Long H<sub>2</sub> plasma treatment

ımec

# CONCLUSION

- ALD Ru selectivity is achieved with sel. factor 600 (3.2 nm of Ru on the SiCN)
- ALD Ru is deposited selectively into aC trench on the SiCN surface using additional 10s H<sub>2</sub> plasma
- ELD Cu growths bottom-up, but defectivity should be improved



# mec

embracing a better life

